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[Modern Chemistry Chapter 13 Review](#)

Chemistry Chapter 5 Review. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. brianna-griffin. Terms in this set (60) Properties . Mendeleev attempted to organize the chemical elements based on their. Periodic Table. Mendeleev is credited with developing the first successful. Properties. Mendeleev did not always list elements in his periodic table in order of ...

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In the modern periodic table, the electronegativity of elements increases across a period (row) and decreases down a group (column). Therefore, the bottom-left most element (francium) is predicted to have the lowest electronegativity and the top-right most element (fluorine) is predicted to have the highest electronegativity. The noble gases are not considered since they are relatively inert).

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The chemistry of carbohydrates most closely resembles that of alcohol, aldehyde, and ketone functional groups. As a result, the modern definition of a CARBOHYDRATE is that the compounds are polyhydroxy aldehydes or ketones. The chemistry of carbohydrates is complicated by the fact that there is a functional group (alcohol) on almost every ...

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This chapter on the chemistry of milk therefore begins with a brief review of some basic chemical concepts. Chemical symbols of some common elements in organic matter: C Carbon Cl Chlorine H Hydrogen I Iodine K Potassium N Nitrogen Na Sodium O Oxygen P Phosphorus S Sulphur. Basic chemical concepts Atoms. The atom is the smallest building block of all matter in nature and cannot be divided ...

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